

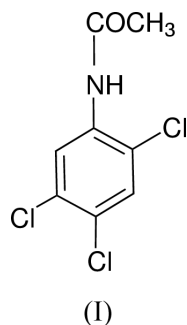
2,4,5-Trichloroacetanilide

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Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
 R factor = 0.023
 wR factor = 0.062
Data-to-parameter ratio = 16.4For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.The title compound, $\text{C}_8\text{H}_6\text{Cl}_3\text{NO}$, also known as *N*-(2,4,5-trichlorophenyl)acetamide, is an organic non-linear optical material. It crystallizes in the monoclinic system, in the non-centrosymmetric space group *Pn*.

Comment

The title compound, (I), is found to be of interest as an organic non-linear optical material. The crystals show optical second harmonic generation with the fundamental beam ($\lambda = 1064$ nm) of an Nd-YAG laser. The single-crystal structure elucidation in a non-centrosymmetric space group further reinforces this observation.Fig. 1 shows the title compound. The torsion angle about the $\text{C1}-\text{N1}$ bond is $40.3(3)^\circ$, which shows that the amide group deviates markedly from the plane of the phenyl ring (Table 1). The packing of molecules is stabilized by intermolecular $\text{N}-\text{H} \cdots \text{O}$ hydrogen bonds, leading to the formation of molecular chains running along the *a* axis (Fig. 2 and Table 2).

Experimental

The title compound was prepared by the direct reaction of 2,4,5-trichloroaniline and acetic anhydride at room temperature for 10 min. Crystals suitable for single-crystal diffraction study were grown at ambient temperature by slow evaporation of a methanol solution. The title compound crystallizes as colourless needles.

Crystal data

 $\text{C}_8\text{H}_6\text{Cl}_3\text{NO}$
 $M_r = 238.49$
Monoclinic, *Pn*
 $a = 3.9015(8)$ Å
 $b = 12.658(3)$ Å
 $c = 9.6687(19)$ Å
 $\beta = 101.186(5)^\circ$
 $V = 468.42(17)$ Å³
 $Z = 2$ $D_x = 1.691$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 825
reflections
 $\theta = 3.9-27.7^\circ$
 $\mu = 0.93$ mm⁻¹
 $T = 293(2)$ K
Needle, colourless
 $0.56 \times 0.28 \times 0.27$ mm

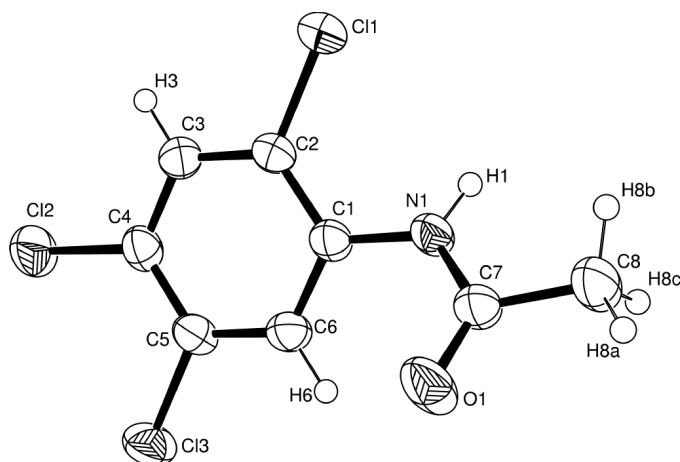


Figure 1
View of the title compound, with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

Data collection

Bruker SMART CCD area-detector diffractometer	1956 independent reflections
φ and ω scans	1903 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.014$
$T_{\text{min}} = 0.624$, $T_{\text{max}} = 0.787$	$\theta_{\text{max}} = 28.0^\circ$
3931 measured reflections	$h = -5 \rightarrow 5$
	$k = -16 \rightarrow 16$
	$l = -12 \rightarrow 11$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0438P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.023$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.062$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.05$	$\Delta\rho_{\text{max}} = 0.18 \text{ e } \text{\AA}^{-3}$
1956 reflections	$\Delta\rho_{\text{min}} = -0.20 \text{ e } \text{\AA}^{-3}$
119 parameters	Absolute structure: (Flack, 1983),
H-atom parameters constrained	817 Friedel pairs
	Flack parameter = 0.04 (4)

Table 1
Selected geometric parameters (\AA , $^\circ$).

C1—C2	1.7277 (17)	O1—C7	1.210 (2)
C2—C4	1.7256 (16)	N1—C1	1.4067 (19)
C3—C5	1.7237 (17)	N1—C7	1.357 (2)
C1—N1—C7	124.18 (14)	C12—C4—C5	121.16 (13)
N1—C1—C6	121.22 (14)	C13—C5—C4	120.95 (13)
N1—C1—C2	120.58 (14)	C13—C5—C6	118.84 (13)
C1—C2—C3	118.11 (12)	N1—C7—C8	114.71 (15)
C1—C2—C1	120.05 (12)	O1—C7—N1	122.92 (16)
C2—C4—C3	118.77 (12)	O1—C7—C8	122.37 (16)
C7—N1—C1—C2	140.98 (18)	C7—N1—C1—C6	-40.3 (3)

Table 2
Hydrogen-bonding geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
N1—H1 \cdots O1 ⁱ	0.86	2.19	3.022 (2)	164
C6—H6 \cdots O1	0.93	2.46	2.865 (2)	107

Symmetry code: (i) $x - \frac{1}{2}, -y, z - \frac{1}{2}$.

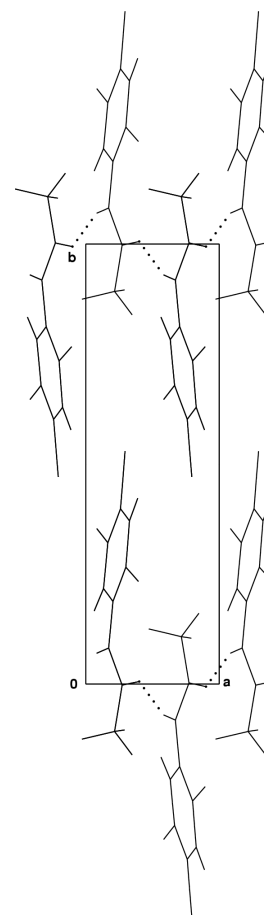


Figure 2
Packing diagram of the title compound, viewed down the c axis.

H atoms were refined as riding, with C—H distances of 0.93 or 0.96 \AA and an N—H distance of 0.86 \AA .

Data collection: *SMART* (Bruker, 1998); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1998); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* for Windows (Farrugia, 1997) and *CAMERON* (Watkin *et al.*, 1996); software used to prepare material for publication: *PLATON* (Spek, 1990).

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References

- Bruker (1998). *SMART* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXL97* and *SHELXS97*. University of Göttingen, Germany.
- Spek, A. L. (1990). *Acta Cryst.* **A46**, C-34.
- Watkin, D. J., Prout, C. K. & Pearce, L. J. (1996). *CAMERON*. Chemical Crystallography Laboratory, University of Oxford, England.